

Light generator: synchrotron

- A synchrotron is an accelerator of electrons. The electrons are maintained in a circular ring by magnetic field and produce light tangentially to their trajectory.
- The light is used to study different systems; applications from medicine to hard physics.

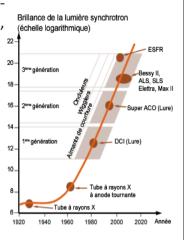






Synchrotrons

- 1st generation: parasitic on highenergy physics operations (DESY, SPEAR, NINA, VEPP)
- 2nd generation: Dedicated X-ray sources (SRS, CHESS, NSLS, Photon Factory)
- 3rd generation: High Brilliance, use of insertion elements: undulators and wigglers (ESRF, APS, MAX2, SLS, DIAMOND, SOLEIL)
- 4th generation: Free electron Lasers



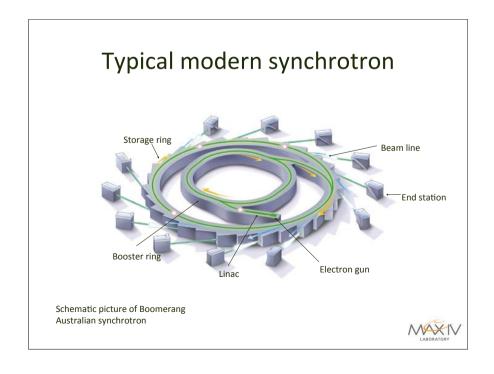


Properties of synchrotron light

- **High brightness:** synchrotron light is extremely intense and highly collimated.
- Wide energy spectrum: light is emitted with energies ranging from infrared light to hard, energetic (short wavelength) X-rays.
- Tunable: through sophisticated monochromators and insertion devices, it is possible to obtain an intense beam of any selected wavelength.
- **Highly polarised:** the synchrotron emits highly polarised radiation, which can be linear, circular or elliptical.







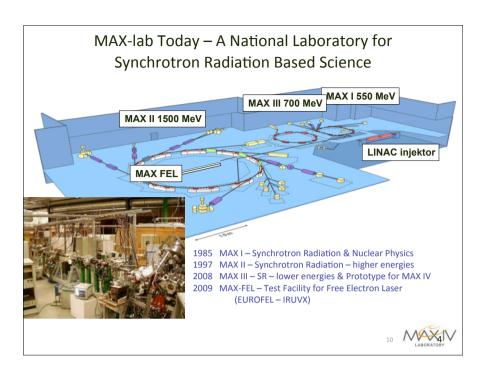
MAX-lab, One Out of Two Swedish National Laboratories • Host: Lund University Operated by: LU & Swedish Research Council (VR) • National laboratory – "open access" Other Countries Norway Finland Globborg University of Agricultural Sciences Special University Chainers University of Technology University College + Commercial users

Some history on biology at synchrotrons

- Natural Source for "synchrotron radiation": Space
- First synchrotron radiation,1947
- X-ray spectra, 1963 at Frascati synchrotron
- 1971, first X-ray-diffraction experiments at DESY (Rosenbaum & Holmes)
- 1975: First protein diffraction at SPEAR
- During 70's MAD, EXAFS, SAXS, DNA Fiber diffraction pioneered
- At the end of the 1980's ESRF and APS proposed



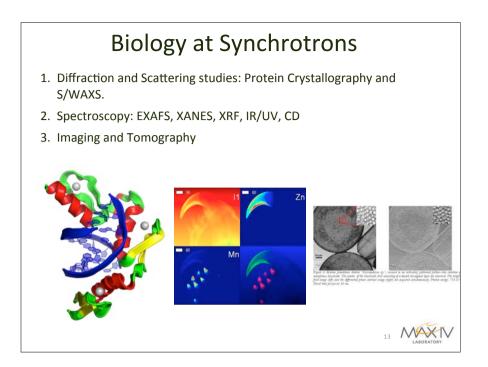


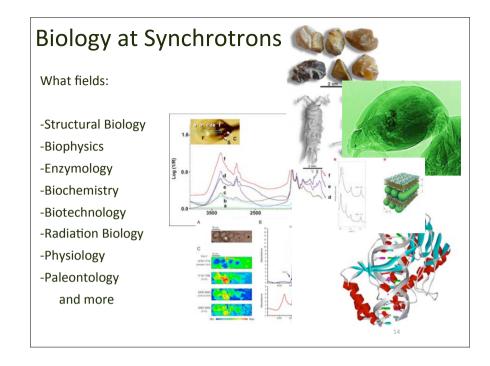


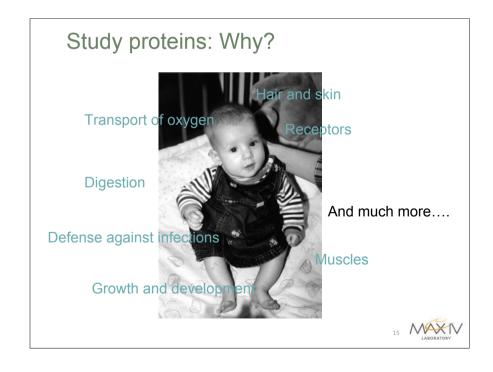
Biology at synchrotrons, why

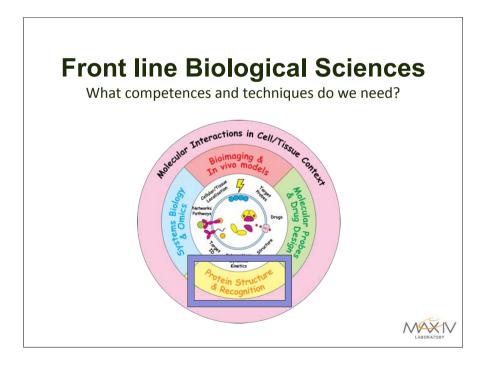
- Plusses:
- High intensity
- Small focus
- Higher flux at the sample
- Range of wavelengths available
- Other resources available (experts, lasers, labs etc.)
- Minusses
- Samples get destroyed: Radiation damage
- One has to travel to the synchrotrons











Methods to study 3D structures.

- 1. Electron microscopy.

 No size limit, but particles need to be big. Medium resolution
- 2. NMR. >40 000 Da, high solubility
- 3. X-ray Crystallography. Crystals! No size limit, high resolution.
- Modelling
 Extensively used but of "limited" value.

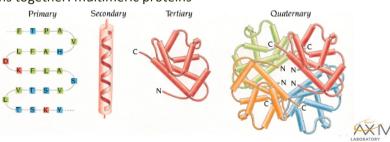
Structure in four dimensions:

Primary Structure Amino acid sequence.

Secondary Structure Local regular structure: α -helices and β -sheets.

Tertiary Structure Packing of secundary structure into one or several compact globular domains

Quarternary Structure The arrangement of several folded chains together: multimeric proteins



Proteins are polymers

Proteins are formed by a chain of repeating molecules. One such molecule is called an amino-acid. There are 20 types of amino-acids but they have all a common backbone or main-chain:



The protein chain is formed by linking the amino-acids together. The linkage is called the peptide bond:



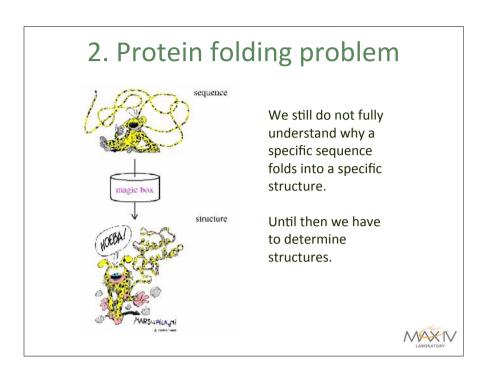
The chain of amino-acids linked to each other by peptide bonds is also called: polypeptide chain.

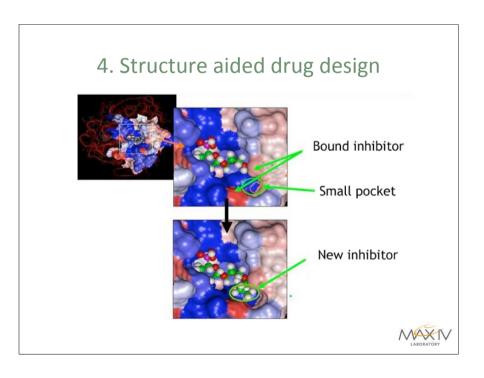
Why study structure? Java DNA Active Center Clamp RNA Exit Groupes

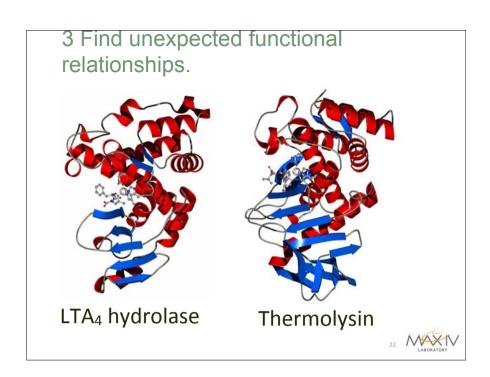
1. FUNCTION IS STRUCTURE!

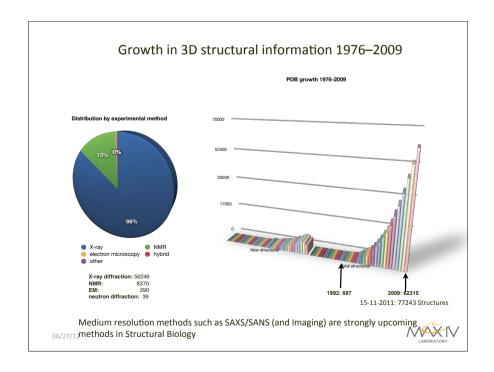
Structure of RNA polymerase: clues of how it reads DNA and makes RNA



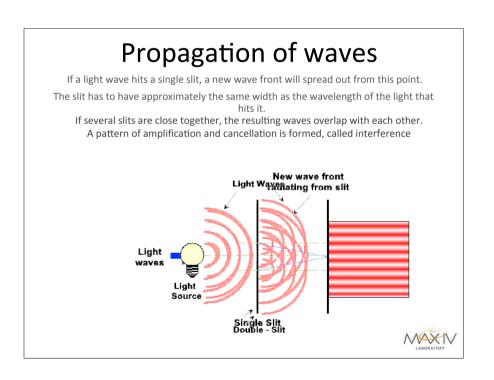


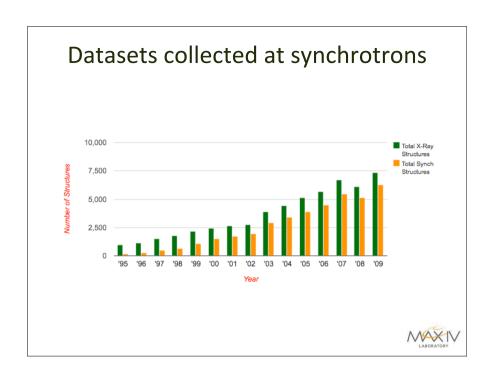


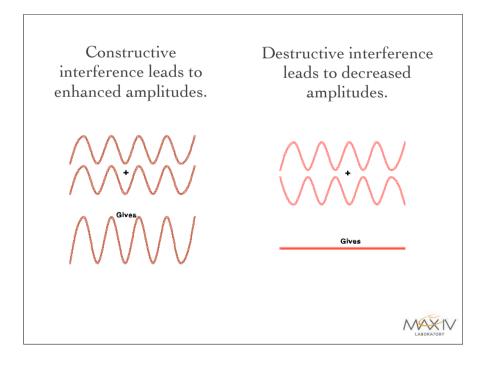




Reasons explosion 1. Molecular Biology developments 2. Computer developments 3. Synchrotrons 1. Molecular Biology India Serie, Selar India Mon. 38 GHz. Appa 2164.0 3 GHz. Appa 216







X-ray diffraction

To be able to see certain details we need light with approximately the same wavelength as what we want to see.

Atomic structure has details with distances on the order of 1-2 Å (1Å=10⁻⁷m)

Hard X-ray radiation has wavelengths around 1Å.

Electrons and neutrons have similar wavelengths and can also be used for diffraction and scattering experiments.



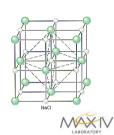
Why crystals?

One molecule scatters X-rays poorly. Many molecules are needed. If they are organized in a crystal it simplifies work tremendously.

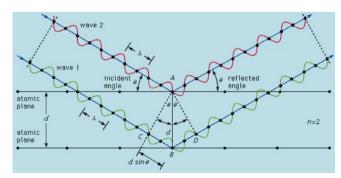
Bragg (father and son, Nobel price 1915) realized that the atomic structure of a molecule could be established using X-ray diffraction on crystals

X-ray diffraction can be related to microscopy.





Basis for structure determination by diffraction: Bragg's law



For constructive diffraction waves need to be in phase (λ) so CB + BD = n (λ) ==> 2dsin θ =n(λ) θ is called the glancing angle Measured reflex = F(hkl)

MAX V

Electron density

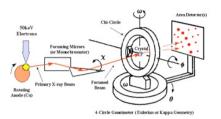
If measure all reflections from all planes F(hkl) we can calculate the electron density distribution:

 $\rho(r) = 1/V \sum F_{hkl} \exp(-2\pi i(hx+ky+lz))$ V = volume of the unit cell

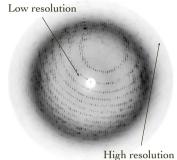
This is a Fourier synthesis.



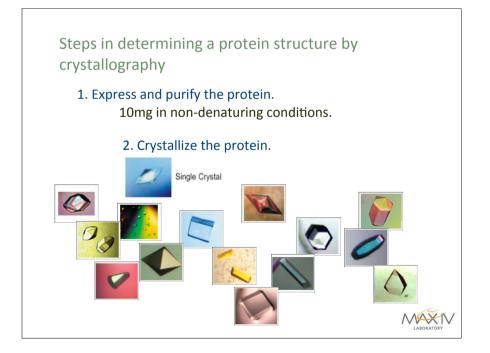
How does this relate to experiment?

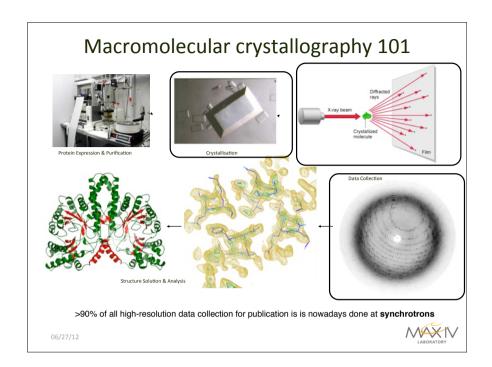


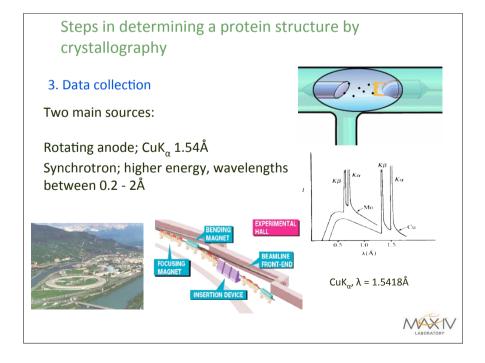
Crystal rotated by about 1° Many pictures collected to get full "reciprocal space"



Several hundred planes went through the diffracting position during 1° Each one has unique indices h,k,l







Importance of synchrotron radiation

High intensity SR improves resolution limits

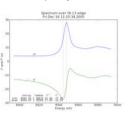
Highly intense beams with low divergence allow the study of smaller samples and larger complexes

Difficult targets (e.g. membrane proteins) may give only microcrystals

Accurately tuneable SR is essential for modern methods to solve the "phase problem", e.g. multiple wavelength anomalous dispersion

The high-throughput methods require intense radiation





Typical anomalous edge

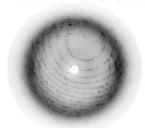


Steps in determining a protein structure by crystallography

4) Solve the phase problem

$$\rho_{(x,y,z)} = \frac{1}{V} \sum_{k} \sum_{l} \sum_{k} F_{(h,k,l)} \exp[-2\pi \cdot i(hx + ky + lz)]$$

$$\rho_{(x,y,z)} = \frac{1}{V} \sum_{k} \sum_{l} \sum_{l} |F_{(h,k,l)}| \exp[-2\pi \cdot i(hx + ky + lz - \alpha_{(h,k,l)})]$$



Not straightforward!



Diffraction pattern:

Electron density

What is the phase problem?

$$\rho_{(x,y,z)} = \frac{1}{V} \sum_{h} \sum_{k} \sum_{l} F_{(h,k,l)} \exp[-2\pi \cdot i(hx + ky + lz)]$$

$$\rho_{(x,y,z)} = \frac{1}{V} \sum_{h} \sum_{k} \sum_{l} |F_{(h,k,l)}| \exp[-2\pi \cdot i(hx + ky + lz - \alpha_{(h,k,l)})]$$



How to overcome the phase problem?

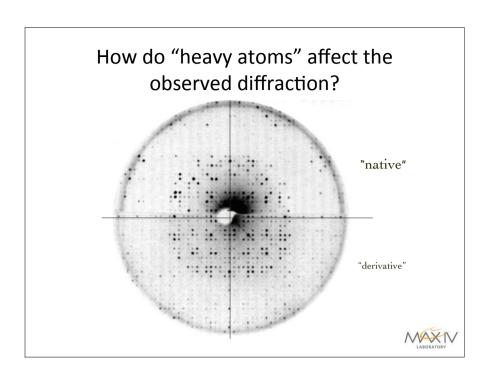
Different methods to overcome the phase problem:

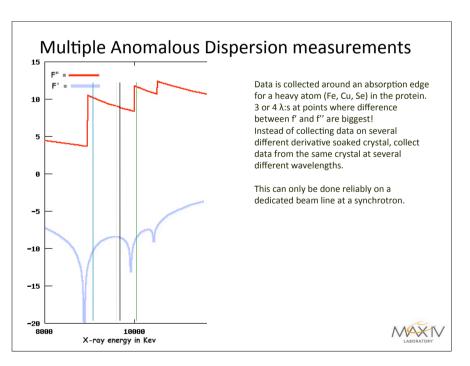
- -Patterson methods
- -Direct methods

In protein crystallography:

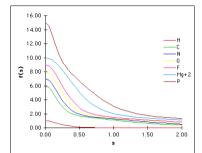
- -Multiple isomorphous replacement(MIR)
- -Single or Multiple anomalous dispersion measurements (SAD or MAD)
- -Molecular Replacement (MR)







What is anomalous dispersion?



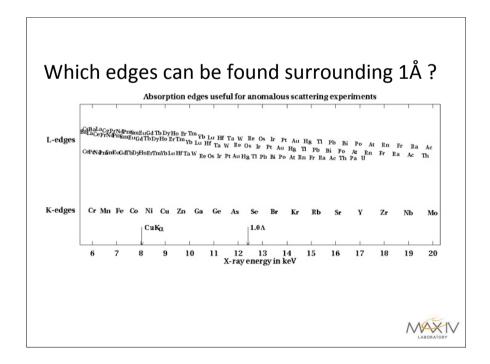
Each element has its own scattering factor f. Scattering factors ~ amount of electrons/resolution

$$f^{o}(\sin\theta/\lambda) = \sum_{i=1}^{4} a_{i} \cdot e^{-b_{i}(\sin\theta/\lambda)^{2}} + c$$

Near absorption edges of elements f gets modified:

$$f = f^o + f' + i \cdot f''$$

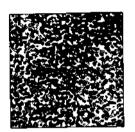


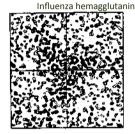


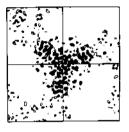
Steps in determining a protein structure by crystallography

5. Improve electron density (phase angles)

Local symmetry averaging (e.g. viruses), Solvent flattening. The solvent regions (more than 50% of the crystal) should have a flat density.







Experimental map: single heavy atom derivative

One cycle of averaging

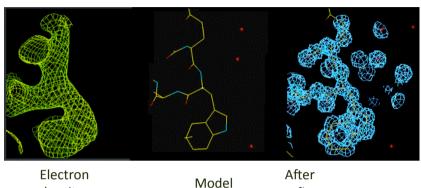
Eleven cycles of averaging

2009 Nobel price for chemistry

Steps in determining a protein structure by crystallography

6. Interpret the electron density.

This is now done with more and more automatic methods.



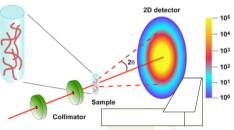
density map

refinement

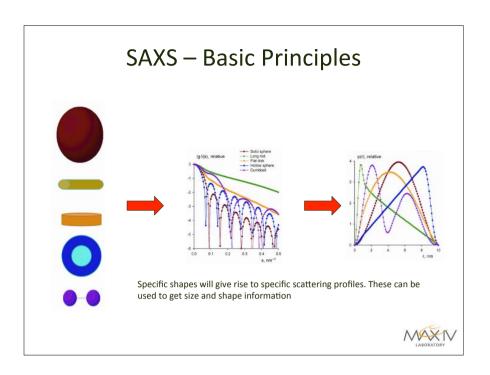


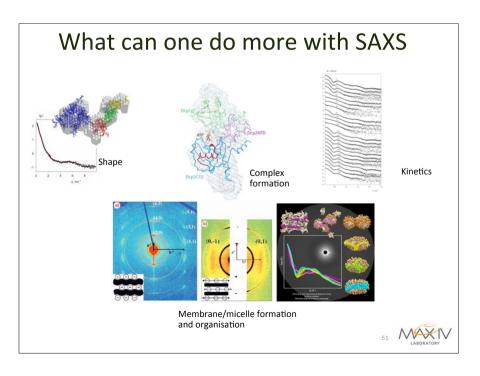
Small angle scattering of X-rays (SAXS)

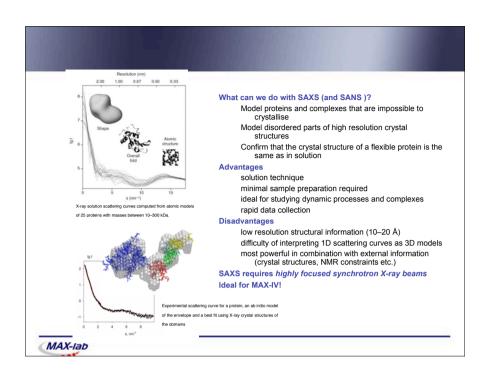
- Used since the 1960's to obtain low resolution structural information in the absence of crystals.
- Possible to obtain information on domain organisation and detailed modeling of macromolecular complexes using rigid body refinement.

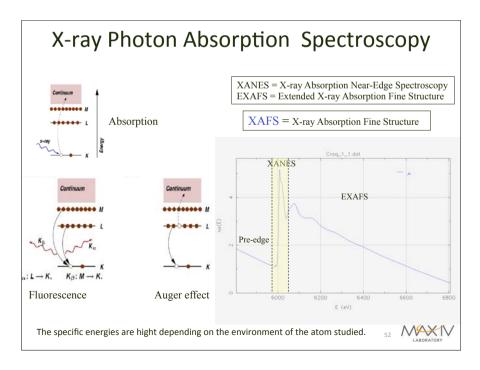




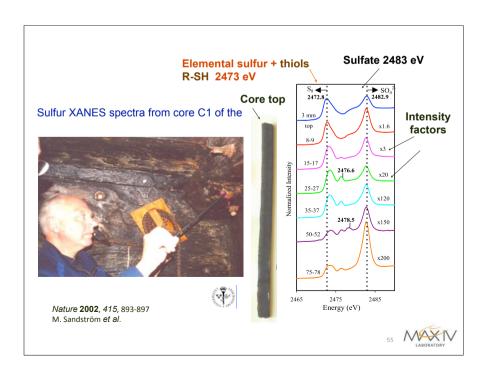






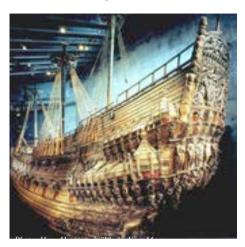


Spectroscopy:XAS (EXAFS, XANES), XRF • Elegant probe for element speciation in biological and environmental samples. Probes can be very small: microns. • Element-specific structural information. • Detailed information on metal centers in proteins. Jano et al., Science 2006 Bergmann U et al. PNAS 2010;107:9060-9065 Wolfe-Simon et al., Science 2010



What is happening with the Vasa ship?

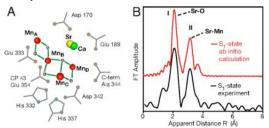






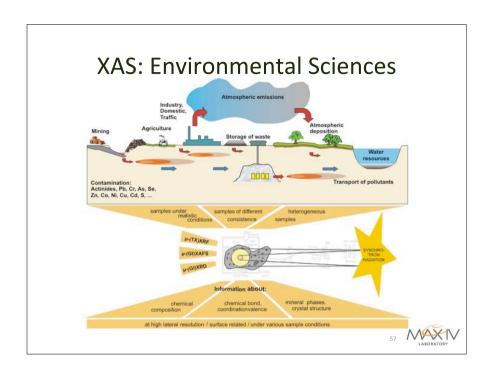
EXAFS looking at metallocenters in proteins

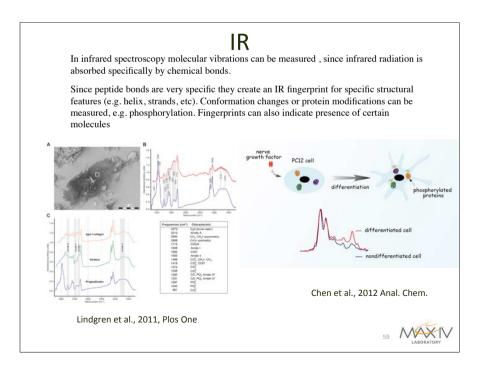
- · Metallocenters in proteins are often very important for the function they perform
- Often involved in catalysis, especially transition metals
- EXAFS probes a specific element
- can reveals the oxidation states
- give detailed information of the metal centre coordination (distances precise within 0.01 Å).
- can be done on crystals but also in solution

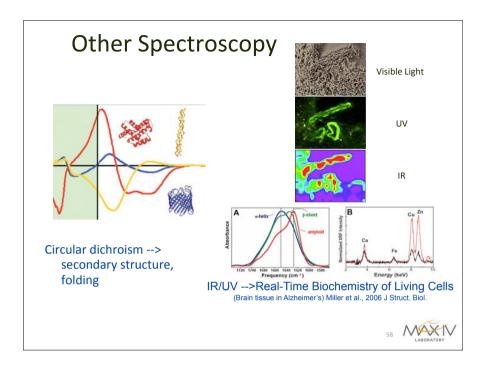


Metal centre of PS-II studied with combination of X-ray diffraction and EXAFS



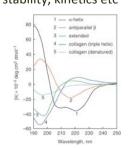


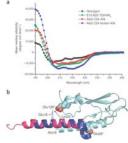




Circular Dichroism

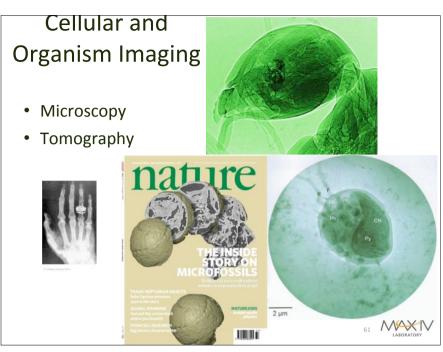
- Chiral molecules absorb left and right polarised light differently.
- This is what is being exploited in Circular Dichroism, in the UV, secondary structures of proteins can be probed.
- SR light gives access to lower wavelengths and better signal to noise.
- Used to study secondary structure contend of proteins, folding, stability, kinetics etc







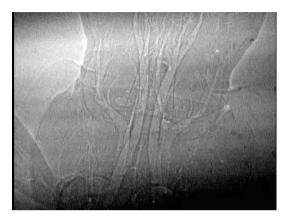
Day et al., 2009 Nat. Chem. Biol.





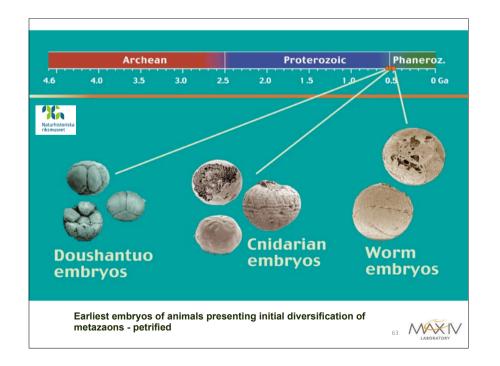
Imaging at synchrotrons

- Possibility to study live animals (insects) to look physiological processes, morphology of organs, penetrating the opaque shell.
- Processes like breathing, feeding, locomotion and more have been studied

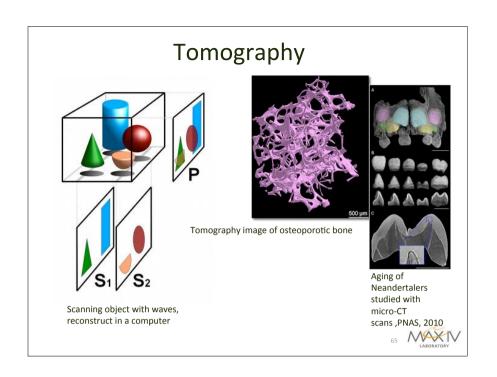


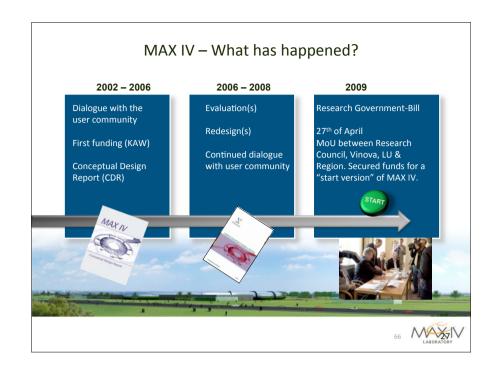
Westneat et al., Science 299 (5606) 558-560.





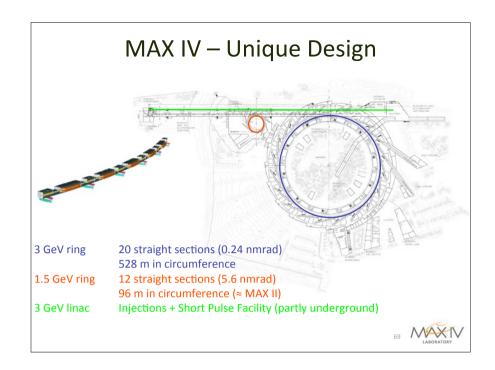


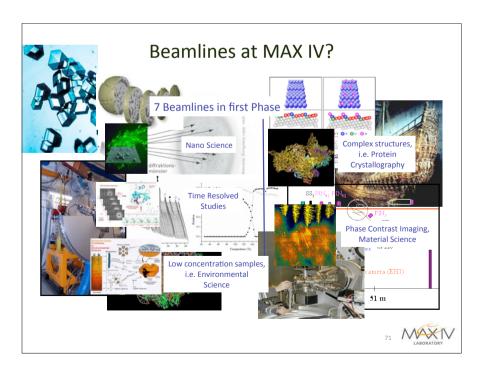


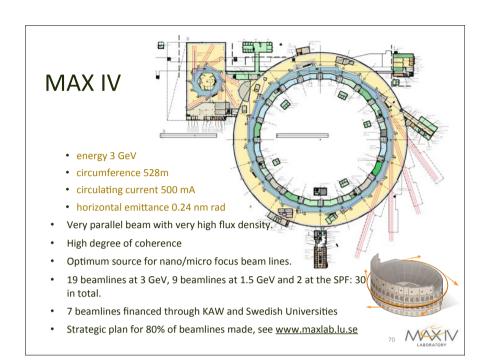














Needs and Possibilities for

For macromolecular crystallography- MAX IV Laboratories

- More brilliant, more focused beam for micro-crystals (e.g. membrane proteins)
 - MAX-II >150 μM; MAX-IV < 1 μM
 - Beam size and focus of MAX IV will be the best in the world!
- Challenging studies
 - Large protein complexes small crystals, large unit cells
 - Membrane proteins small crystals
- Automation and high throughput
 - few minutes vs. several hours per data set
 - rapid data collection for kinetic studies
 - up to 1000 samples per day?
- Combination with spectroscopic methods (UV/vis. EXAFS etc.)

For macromolecular crystallography- ESS

Higher Intensity source with detectors such that work on moderate unit cells will be possible

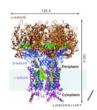
For solution studies

- A dedicated SAXS beam line
- A dedicated SANS beam line

Future

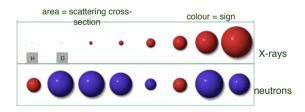
Imaging beam lines





10 um crystals

Neutrons and X-rays are complementary



Increasing atomic number

- X-rays interact with the electron clouds around atoms
- · Strength of interaction depends on the no. of electrons
- Neutrons interact with the nuclei
- No linear relationship with atomic number
- · Neutrons also have a magnetic moment
- Neutrons are non-destructive

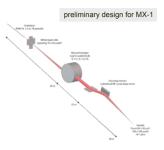


Plans for structural biology at MAX IV

- Macromolecular crystallography
- Two beamlines proposed, both tuneable wavelength and with in situ spectroscopy
- . BIOMAX (First Phase)
 - High throughput, highly automated beamline
 - Beam size at crystal 20-100 μm
 - Tried and tested technology
 - Operational and reliable from day 1
 - Sample changer robotics
 - Remote access
- MX-2
 - Microfocus heamline
 - Cutting edge technology
 - Beam size at crystal < 1–5 μm

Small-angle X-ray scattering

- A SAXS beamline (both bio and non-bio)
- Reduced sample volumes (from 100s of μ l to sub- μ l using microfluidics
- In a longer perspective: Biological/Medical Imaging





www.bioxtas.org



Why spallation sources?

- · The traditional source of neutrons is a nuclear reactor
- Very large and weak beam compared to X-rays
- An diffraction experiment that takes 5-10 minutes with X-rays at the ESRF could take several days with neutrons at the nearby ILL.
- Only three other spallation sources in the world
 - Spallation Neutron Source (Oak Ridge National Laboratory, Kentucky, USA)
 - Los Alamos National Laboratory
 - J-PARC (Ibaraki, Japan)



Neutron Spallation Permay Perdon A high velocity proton is accelerated into a heavy particle (e.g. Hg nucleus). A number of spallation particles is produced amongst which neutrons

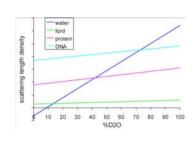
the hydrogen atoms proposed reaction mechanism for aspartic proteases transition state stabilised by negative charge on Asp32 studies on human aldose reductase: 0.66 A X-ray map: 54% of H visible perdeuteration is very important endothiapepsin complexed with gem-diol transition-state mimicking inhibitor Neutron map at 2.0 Å resolution reveals the protonation states An X-ray map at 1.0 Å resolution did not reveal any protons Coates et al. (2008), JACS 130, 7235

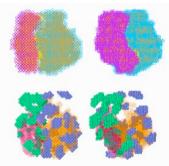
Biology at a Neutron Source

- Many techniques same as those that are used at a synchrotron
- Crystallography
- Small Angle Scattering
- Reflectometry
- Spectroscopy
- Imaging



Scattering: Neutrons give contrast





- By playing with the level of D2O/deuteration of components in an experiment, specific components can be highlighted or blanked out.
- Especially in studies of complexes this can be a big advantage.

The future is bright.....



